Cucurbitacins – An insight into medicinal leads from nature

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ABSTRACT

Cucurbitacins which are structurally diverse triterpenes found in the members of Cucurbitaceae and several other plant families possess immense pharmacological potential. This diverse group of compounds may prove to be important lead molecules for future research. Research focused on these unattended medicinal leads from the nature can prove to be of immense significance in generating scientifically validated data with regard to their efficacy and possible role in various diseases. This review is aimed to provide an insight into the chemical nature and medicinal potential of these compounds exploring their proposed mode of action, probable molecular targets and to have an outlook on future directions of their use as medicinal agents.

Key words: Cucurbitaceae, cucurbitacin, triterpenoids

INTRODUCTION

Plant secondary metabolites represent tremendous resources for scientific and clinical research as well as for new drug development. Cucurbitacins are multiplex category of diverse compounds found in the plants of family Cucurbitaceae. Medicinal and toxic properties of these compounds have stimulated a continuing interest in them.^[1] Many genus of Cucurbits viz. Trichosanthes, Cucurbita, Cucumis and Citrullus are affluent in cucurbitacins. These compounds have also been discovered in other plant families like Scrophulariaceae, Cruciferae, Datiscaceae, Primulaceae, Rubiaceae etc., The diversity of cucurbitacins lies in variety of its side chain derivatives that contribute to their disparate pharmacological actions.^[2,3] The bitter taste of plant species like cucumber have been attributed to the presence of cucurbitacins. The first cucurbitacin was isolated as a crystalline substance in 1831 and was named α -elaterin. Certain plant species rich in cucurbitacins like Momordica hold coveted position in different system of traditional medicines for curative effects in metabolic

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disease like diabetes. Plants from genus *Trichosanthes* have been used in China by herbal drug practitioners.^[4] The purpose of this review is to gather the information related to these highly diverse group of compounds which may be useful in future research.

OCCURRENCE

Cucurbitacins are found in many cucurbitaceous plants. They are most common in species of the Bryonia, Cucumis, Cucurbita, Luffa, Echinocystis, Lagenaria and Citrullus. The plants of genera Momordica contain a special group of Cucurbitacins called momordicosides. The level of Cucurbitacins varies between tissues. They may be concentrated in fruits and roots of mature plants. In fruits where Cucurbitacins are produced, their highest concentration is achieved on maturity. Seeds generally contain very low concentration of Cucurbitacins. Cucurbitacin producing plants have also been identified outside the cucurbitaceae in the members of Scrophulariaceae, Begoniaceae, Primulaceae, Liliaceae, Tropaeolaceae and Rosaceae. The seeds of certain cruciferous plants, like Iberis species and Lepidium sativum also contain cucurbitacins.^[4] It is reported that Cucurbitacins are formed in situ and are not transported to other parts of the plant.^[5] The distribution of Cucurbitacins among various families of plant kingdom has been depicted in [Figure 1].

CHEMISTRY AND CATEGORIES OF CUCURBITACINS

All cucurbitacins contain a basic 19- $(10 \rightarrow 9\beta)$ -abeo--10 α -lanost-5--ene ring skeleton. A common feature among all compounds in the category of Cucurbitacins is the presence of 5,(6)--double bond. The difference of Cucurbitacins from steroidal nucleus lies in the fact that in basic structure of Cucurbitacins a methyl group is located at C-9 rather than C-10.^[6] Most of the Cucurbitacins are tetracyclic, but some representatives have an extra ring due to formal cyclization between C--16 and C--24 as in cucurbitacins S and T.^[7] The Cucurbitacins differ from most of the other tetracyclic triterpenes by being highly unsaturated and contains numerous keto--, hydroxyl--, and acetoxy--groups.^[8] Certain Cucurbitacins have been discovered in the form of glycosides and some of them lack C--11 carbonyl function.^[9] Chemically, Cucurbitacins are ranked according to presence of various functional groups on rings A and C, diversity in side chain and stereochemical considerations.^[10] The structural composition of following Cucurbitacins are known and have been designated by the letters: A, B, C, D, E, F, G, H, I, J, K, L, O, P, Q, R and S [Figure 2]. The term --""Cucurbitacin""-- refers to group of Cucurbitacins along with their glycosidic forms mentioned above, including those forms listed before.[11] Cucurbitacin G and H have same structures but differ from each other in the configuration of the hydroxyl group at position 24 which is not yet established.^[12] Cucurbitacin R was demonstrated to be 23, 24-dihydrocucurbitacin D (DHCD) hence, its description has been moved to the group of Cucurbitacin D.^[13] Similarly Cucurbitacin J and K differ from each other only in the configuration of hydroxyl group at position 24 which is yet to be determined.^[14] A special group of Cucurbitacins are called as momordicosides, named after their occurrence in Momordica charantia. Momordicosides have never been identified in any other plant species. The common feature of momordicosides is that C₁₀ has been oxidized to an aldehyde group.

IDENTITY, PHYSICAL AND CHEMICAL PROPERTIES

Cucurbitacins are derived from basic cucurbitane ring skeleton which is a triterpene hydrocarbon (IUPAC name 19 (10-9 β)-abeo-5 α -lanostane, which on modification by groups containing oxygen and double bonds produce manifold Cucurbitacins with distinctive features.^[15] The saccharide linkage

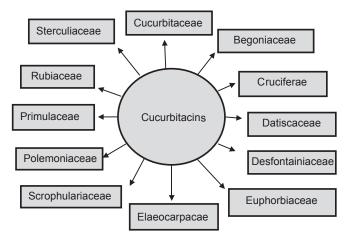


Figure 1: Occurrence of Cucurbitacins in various plant families

is generally present at C-2 (2-O- β -glycosides) in Cucurbitacin glycosides. Majority of Cucurbitacins are usually crystalline in nature or present in the form of needles at room temperature except Cucurbitacin H which is an amorphous solid. Most Cucurbitacins are soluble in petroleum ether, chloroform, benzene, ethyl acetate, methanol and ethanol, but are insoluble in ether. They are only slightly soluble in water. Cucurbitacins usually have absorption maxima for ultraviolet light between 228-234 nm.^[16] The molecular formulae and properties of all the known crystalline Cucurbitacins are given in Table 1.

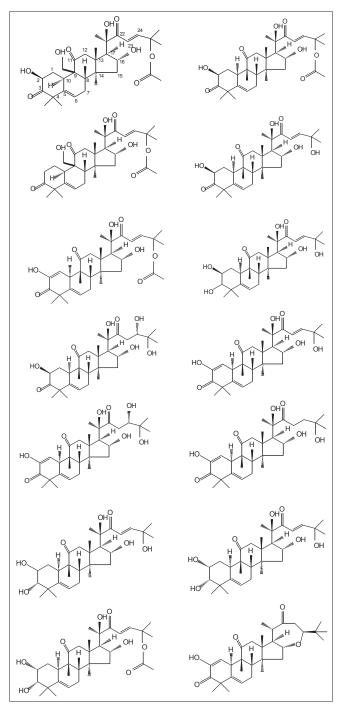


Figure 2: Chemical structures of cucurbitacin analogs

ANALYSIS OF CUCURBITACINS

Cucurbitacins are secreted in very minute quantities in plants and due to difficulties in isolating these compounds in large quantities many of their potential uses remain unexplored. Cucurbitacins are moderately polar compounds and have been isolated using solvents like methanol. The solubility of aglycone portion of most Cucurbitacins is significant in moderately polar solvents like chloroform. Partition between water and chloroform is customarily used for purification of Cucurbitacins isolated from plant extracts which are extracted with methanol. The chromatographic techniques like open-column chromatography on silica gel, alumina or florisil, or thin layer chromatography have been utilized for purification process of Cucurbitacins from plant extracts.^[6,8] Extraction of cucurbitacins have also been tried using maceration of the plant material with absolute ethanol and lead acetate in equal quantities. The mixture is filtered and after aqueous potassium dihydrogen phosphate is added to precipitate lead. The Cucurbitacins are extracted from the aqueous phase with chloroform three times and finally the extract is concentrated at 70° C.^[17]

A confirmatory test to determine the presence of Cucurbitacins in extracts and fractions has been reported whereby the sample is mixed wih triphenyltetrazolium chloride. The occurrence of red precipitate of formazin is indicates presence of Cucurbitacins.^[18] The thin layer chromatographic solvent mixtures reported in literature have been compiled in Table 2.^[17,19-24] Often invariable presence of α , β -unsaturated ketones either in the side chain or in the A-ring of cucurbitane skeleton results in the UV-absorbance at 230 nm for most Cucurbitacins, yet for many other Cucurbitacin analogues UV-absorbance does not go above 210 nm.^[25] Chromatographic system such as reversed phase high performance thin layer chromatography (HPTLC) using mobile phase composition of ethyl acetate and benzene (25:75) have been reported for quantifying Cucurbitacins.^[26] A high-performance liquid chromatography (HPLC) chromatographic technique using gradient elution of acetonitrile in water have been documented for the analysis of a number of Cucurbitacin analogues commonly found in plants.^[19,27] A general scheme for extraction and isolation of cucurbitacins from plants is summarized in Figure 3.

BIOACTIVITY

Various biological activities attributed to Cucurbitacins with probable mechanish of action (s) have been summarized in Table 3.

Anti-inflammatory activity

Cucurcitacin analogues viz. Cucurbitacin R and DHCB have been reported to possess anti-inflammatory potential and their action is reported to be mediated by inhibition of tumor necrosis factors (TNF)- α and other mediators of inflammation such as nitric-oxide synthase-2 and cyclo-oxygenase-2.^[45,46] Cucurbitacins B, D, E and I have been reported to inhibit cyclooxygenase (COX)-2 enzymes with no effect on COX-1 enzymes.^[34] The anti-inflammatory response of 23, 24-dihydrocucurbitacin D (DHCD) have been hypothesized to get mediated through blocking of NF- κ B activation thereby obstructing the release of nitrous oxide. DHCD can be taken up as probable lead and appraised for providing a promising anti-inflammatory agent.^[36,37]

Antitumor activity

Very less information is available on the role of Cucurbitacins at molecular level which has lead to slow advancement in the development of Cucurbitacins as anti-cancer agents.^[4] In relation to cancer, targets of Cucurbitacin actions involve growth inhibition, arrest of cell cycle at G2/M phase and induction of apoptosis in cancer cell.^[47] The mechanisms underlying anti-tumorigenic potentials of Cucurbitacins involve inhibition of Janus kinase/Signal Transducer Activator of Transcription 3 (JAK/STAT3) signaling pathway whose activation is required for the proliferation and sustainment of cells.^[28-30] The role of Cucurbitacin I in suppressing phosphotyrosine STAT3 in cancer cell lines and cancerous lung

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Cucurbitacin	Nature	Formula	UV max. (ethanol) nm	Mass	m.p.
A	Crystals	C ₃₂ H ₄₆ O ₉	229, 290	574.314	207-208°
В	Crystals	C ₃₂ H ₄₆ O ₈	-	558.3192	184-186°
С	Needles	C ₃₂ H ₄₈ O ₈	231, 298	560.3348	207-207.5°
D	Needles	C ₃₀ H ₄₄ O ₇	230	516.3087	151-153°
E	Crystals	C ₃₂ H ₄₄ O ₈	234, 268	556.3035	234.5°
F	Needles	C ₃₀ ³² H ₄₆ ⁴⁴ O ₇	-	518.3243	244-245°
G	Crystals	C ₃₀ H ₅₂ O ₉	-	534.3192	150-152°
Н	Amorphous solid	C ₃₀ H ₄₆ O ₈	-	534.3192	150-152°
1	Needles	C ₃₀ H ₄₂ O ₇	234, 266	514.293	148-148.5°
J	Crystals	C ₃₀ H ₄₄ O ₈	270	532.3036	200-202°
К	Needles	C ₃₀ H ₄₄ O ₈	270	532.3036	200-202°
L	Needles	C ₃₀ ⁴⁴ O ₇	270	516.3087	137-142°
0	-	C ₃₀ H ₄₆ O ₇	-	518.3243	122-127°
Р	-	C ₃₀ H ₄₈ O ₇	-	520.3399	-
Q	-	$C_{32}^{30}H_{48}^{40}O_{8}^{\prime}$	-	560.3348	-
S	-	C ₃₀ ⁴⁰ H ₄₂ O ₆	-	498.298	-

cells of humans has been reported.^[48] Although Cucurbitacin B, E, and I act by inhibiting the activation of both JAK2 and STAT3, Cucurbitacin A and I acts by inhibition of only JAK2 and STAT3 respectively.^[49] It has been reported that Cucurbitacin E inhibited tumor angiogenesis by inhibiting JAK-STAT3 and mitogen

Table 2: List of TLC solvent mixtures reported in literature for cucurbitacins

Solvent system	Solvent ratio	Visualisation				
Toluene: Ethyl acetate	40:60	Vanillin/orthophosphoric acid				
Chloroform: Ethanol	95:5	Vanillin phosphoric acid reagent				
Methanol: Water	55:45	UV 254 nm				
EtOAc-C6H6	75:25	Vanillin-orthophosphate in EtOH				
Ether: Hexane:	70:30:5	UV 254				
Methanol						
Chloroform: Methanol	95:5	UV 254				
TLC=Thin laver chromatography, UV=Ultra violet						

activated protein kinases (MAPK)- signaling pathways.^[31] The role of interference with actin cytoskeleton has been attributed to anti-proliferative effects of Cucurbitacin B and E. The anti-proliferative activities have been correlated directly with the disruption of the F-actin cytoskeleton.^[32] It has been proposed that the combination of Cucurbitacin B with docetaxel may augment the chemotherapeutic effects by suppression STAT3 in patients with laryngeal cancer.^[47] It is expected that cucumber fruits have anti-tumor effects since they have been reported to contain Cucurbitacin C.^[50,51] It has been reported that cucurbitacin B exerts an anticancer effect by inhibiting telomerase via down-regulating both the human telomerase reverse transcriptase and c-Myc expression in breast cancer cells.^[33]

Anti-artherosclerotic activity

There have been reports on Cucurbitacin B and E in glycosidic form to exhibit inhibitory effect on lipid

Table 3: Reported biological activities of cucurbitacins with probable mechanism of action

Activity	Mechanism	Reference
Antitumor activity	inhibition of Janus kinase/Signal Transducer Activator of Transcription 3 (JAK/STAT3) signaling	[28-31]
,	pathway	[32]
	disruption of F-actin cytoskeleton	[33]
	Down-Regulation of the c-Myc/hTERT/Telomerase Pathway and Obstruction of the Cell Cycle	
Anti-inflammatory	inhibit the expression of TNF and proinflammatory mediators such as nitric-oxide synthase-2	[34,35]
2	and cyclooxygenase-2	[36,37]
	inhibition of NO generation through blocking NF- B activation	
Artherosclerosis	Inhibition of lipid-oxidation products	[38-40]
	malonaldehyde (MAD) and 4-hydoxynonenal (4-HNE)	
Blood circulation promoter	Inhibition of Na ⁺ /K ⁺ -ATPase	[41]
Immunosuppressant	By inhibiting expression of surface markers CD69 and CD25 required for activation of lymphocytes	[42]
Antidiabetic	Activation of AMPK pathway (a major regulatory pathway for GLUT4 translocation)	[43,44]

TNF=Tumor necrosis factor, AMPK=5'-adenosine monophosphate-activated protein kinase, GLUT4=Glucose transporter type 4

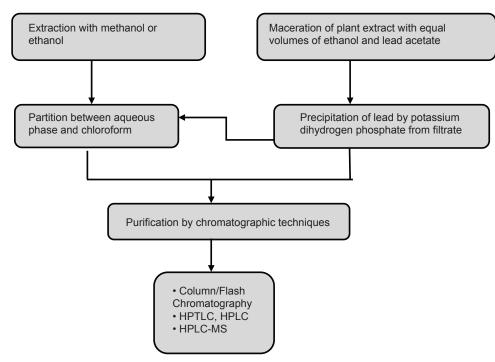


Figure 3: Scheme for extraction and isolation of cucurbitacins. HPTLC= High performance thin layer chromatography; HPLC= High performance liquid chromatography- Mass spectroscopy

oxidation products like- malonaldehyde (MDA) and 4-hydroxynonenal (4-HNE).^[38,39] These reports bolster the therapeutic role of Cucurbitacins in artherosclerosis, which involves modification of lipoproteins by involvement of- MDA and 4-HNE.^[40]

Antidiabetic activity

There have been a plethora of reports on the role of Cucurbitacins for their cytotoxic, hepatoprotective, cardiovascular, and antidiabetic effects.^[36] Cucurbitane triterpenoids present in momordica fruits are noted for antidiabetic and anticancer activities, this may provide leads as a class of therapeutics for diabetes and obesity.[52-55] The 5'-adenosine monophosphate-activated protein kinase (AMPK) pathway is suggested as a probable mechanism for the stimulation of GLUT4 translocation by triterpenoids from M. charantia. It is particularly interesting in relation to diabetes and obesity because activation of AMPK increases fatty acid oxidation, inhibits lipid synthesis, and can improve insulin action.^[43,44] An analogue of 23,24-dihydrocucurbitacin F from Hintonia latiflora has been reported to possess significant hypoglycemic and antihyperglycemic effects. The probable mechanism underlying-- antihyperglycemic effect could be stimulation of insulin release and regulation of hepatic glycogen metabolism.[56]

Miscellaneous activity

It has been reported that the concentration of Cucurbitacin C in the leaves is an important parameter in spider mite resistance in Cucumis sativus, perhaps by acting as an antagonist of a spider mite ecdysteroid receptor.^[57] The steroid like resemblance of Cucurbitacin D may possess therapeutic effects via inhibition of Na⁺/K⁺-ATPase.^[41] The role of Cucurbitacins as preventive and radical scavenging antioxidant has also been reported.^[58] Cucurbitacins have also been reported to possess adaptogenic activity. Cucurbitacins have been reported to increase the rat capillary permeability and to demonstrate antifertility effects in female mice.^[25,59,60] Cucurbitacin D has been reported to inhibit ovulation in mice. There has been protective role of Cucurbitacins acting as allomones in many plant species. Role of Cucurbitacins as anti-feedants for few insects, birds and as kairomones (Cucurbitacin B, E, D, I and L) for diabroticite beetles have been reported.^[11] It is reported that Cucurbitacins act via Cuc receptors located on the maxillary palpi. They arrest the searching behavior of diabroticite beetles and produce a compulsive feeding behavior.^[24] Role of Cucurbitacin B and D in controlling diabrotic beetles can be an interesting approach.[35,61,62]

TOXICITY REPORTS

Substitution pattern on various Cucurbitacins provides the lead to understand and trace out clear distinction between the toxic effects and curative role of Cucurbitacins.^[63] Cucurbitacins have been reported as highly toxic compounds and instances of severe poisoning and death in sheep and cattle that consumed bitter fruits of Cucumis and Cucurbita are well documented.^[64] The range of toxicity of Cucurbitacins based on few in-vivo toxicity reports, has been found to be between 2 -12.5 mg/kg. Although a report on toxicity of Cucurbitacin R at level as high as 375 mg/Kg p.o and 67 mg/ kg i.p is available.^[65] The presence of a double bond at C-23 and acetyl group at C-25 have been found to augment the toxicity of Cucurbitacins.^[66] Cucurbitacin's strong biological activity was found to be very close to their toxic dose, which renders them unlikely to be biological agents.^[48] The extreme bitterness of Cucurbitacins should deter humans from being exposed to substantial quantities of the compounds. Nevertheless, some poisonings have been reported after consumption of Cucurbitaceous food plants.^[8] Cucurbitacins are found to be fatal when fruits of Luffa cylindrical (L.) were consumed.^[67] Gastrointestinal symptoms have also been reported in a Japanese population consuming the bottle gourd, which contained Cucurbitacin D.^[68] The toxicity of Cucurbitacins C, D, E, and I have been assessed and these compounds ascertained to be lethal. Plants with Cucurbitacins C, D, E and I must be avoided as their consumption can lead to illness or even death.^[17] The appearance of toxic symptoms varies with the animal species used in the experiment, the route of administration of the compound, and the quantity that has been administered.[42]

CONCLUSION

Although Cucurbitacins are highly toxic compounds and often their biological activities are close to their toxic dose level, these compounds possess immense pharmacological potential. Apart from their toxic nature cucurbitacins have been proved to possess pharmacological effectiveness against inflammation, cancer, artherosclerosis and diabetes. The reports on their toxicity must not overshadow the potential use of these compounds as potent medicinal agents. The chemical modification of various functional groups of these compounds to reduce toxic effects may provide important lead compounds for future research. Various Cucurbitacin analogues have been explored and are well established for toxic nature and their effectiveness against tumor cell lines. In modern drug discovery from medicinal plants, the importance of Cucurbitaceae species has been markedly recognized in empirical control of diabetes. It is interesting to find most of the traditionally used herbal plants against diabetes especially from genus Momordica are rich in triterpenoids, Cucurbitacins and related compounds momordicosides. Their occurrence is believed to be more in roots and fruits of such plants. The information on absorption, distribution, metabolism and excretion of these compounds is scarce and can be an area of exploration keeping in concern their toxic effects in mammals. Research focused on these unattended medicinal leads from the nature may prove to be of immense significance in generating scientifically validated data regarding their efficacy.

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